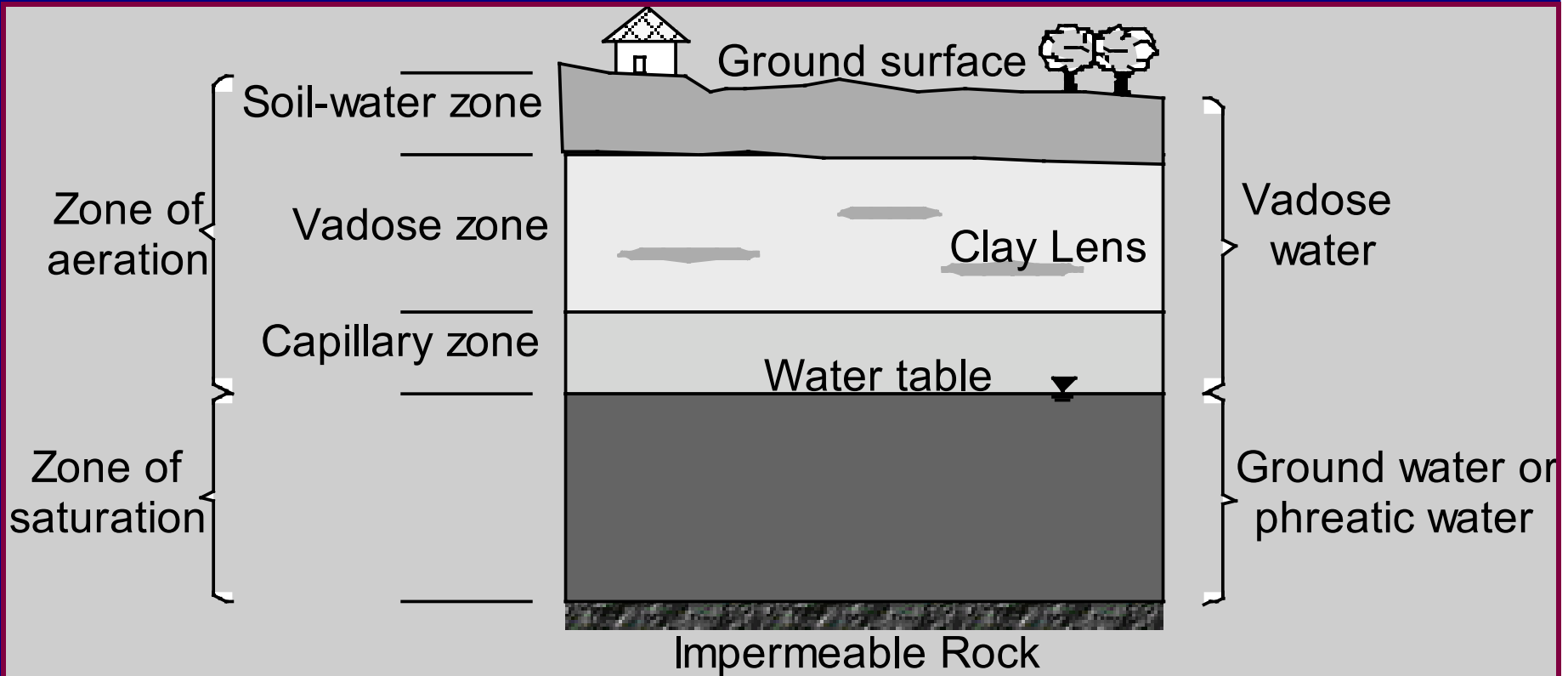


# Ground Water Flow & Contamination: Introduction

**Amended by Dr. Fuentes**  
**after Original by Dr. Philip B. Bedient**  
*(+2005)*

# Vertical Distribution of GW

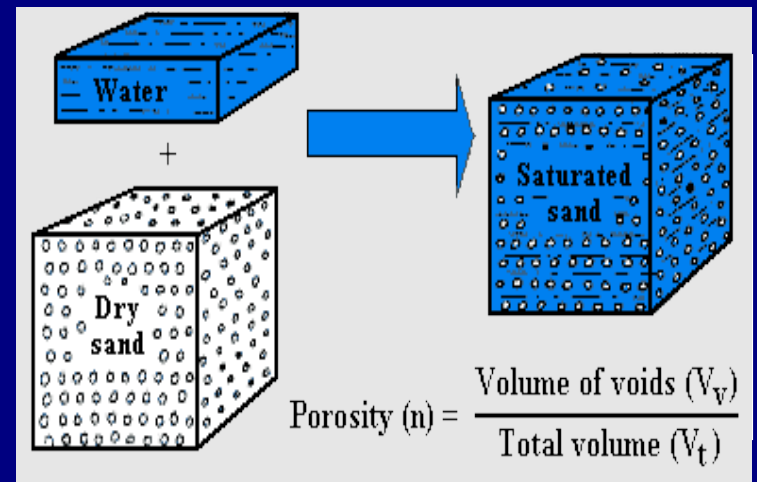
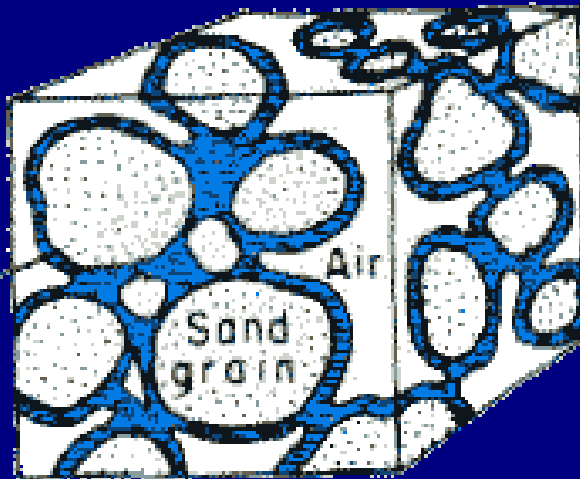


# *Vertical Zones of Subsurface Water*

- Soil water zone: extends from the ground surface down through the major root zone, varies with soil type and vegetation but is usually a few feet in thickness
- Vadose zone (unsaturated zone): extends from the surface to the water table through the root zone, intermediate zone, and the capillary zone
- Capillary zone: extends from the water table up to the limit of capillary rise, which varies inversely with the pore size of the soil and directly with the surface tension

# Porosity

Water



# Soil Classification Based on Particle Size

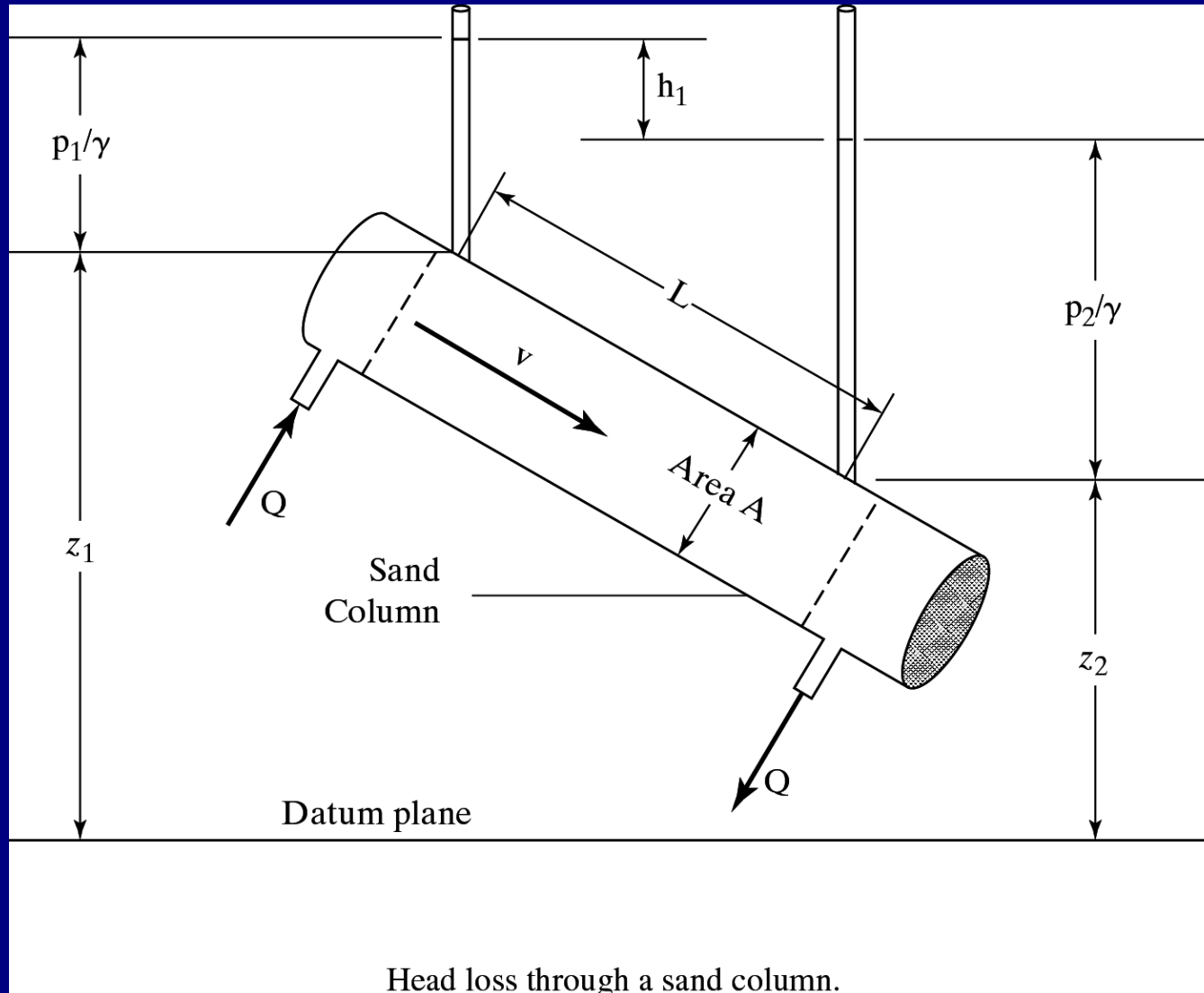
(after Morris and Johnson)

Material	Particle Size, mm
Clay	<0.004
Silt	0.004 - 0.062
Very fine sand	0.062 - 0.125
Fine sand	0.125 - 0.25
Medium sand	0.25 - 0.5
Coarse sand	0.5 - 1.0

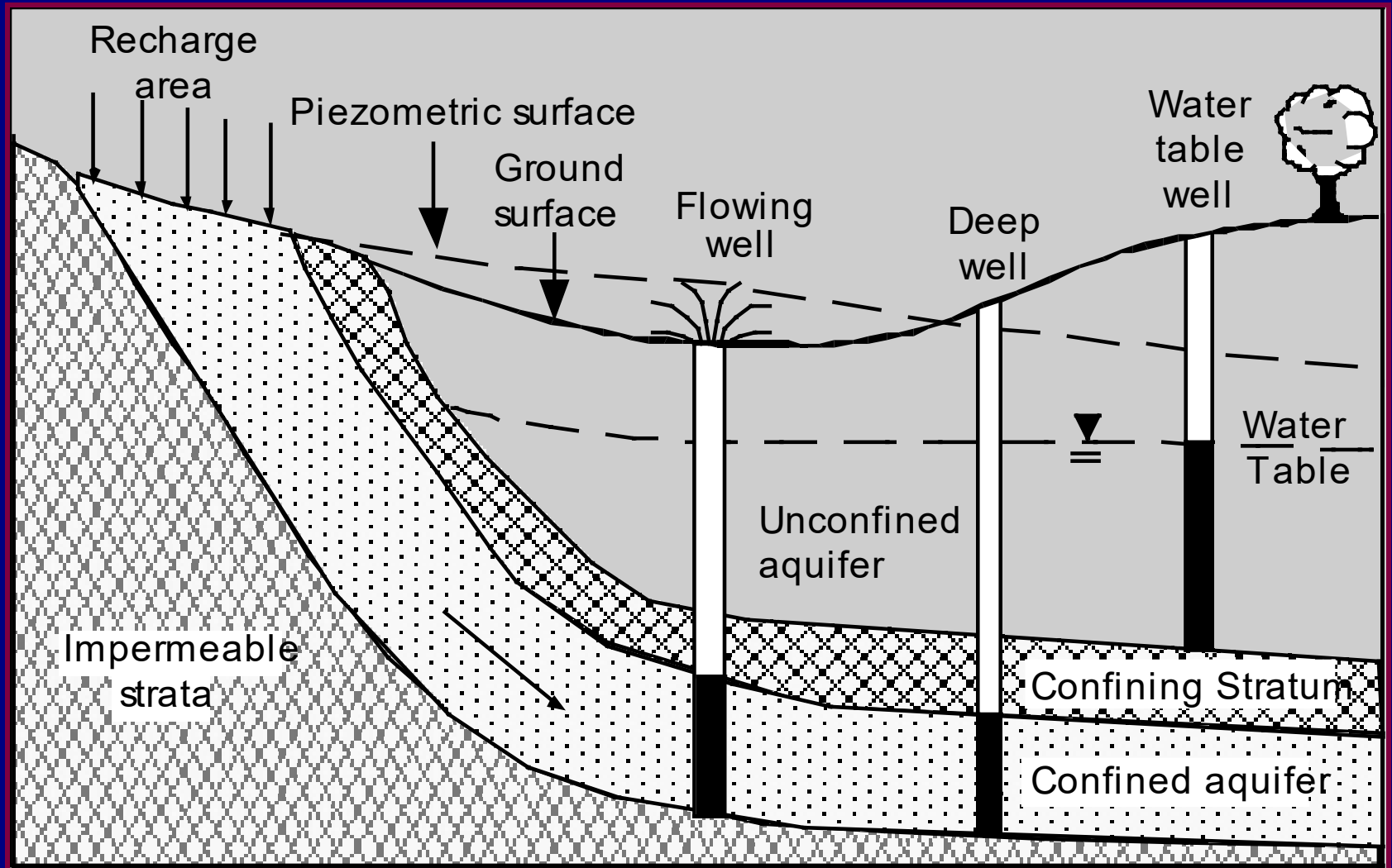
# Darcy's Law

$$V = -K \, dh/dl$$

$$Q = -KA \, dh/dl$$



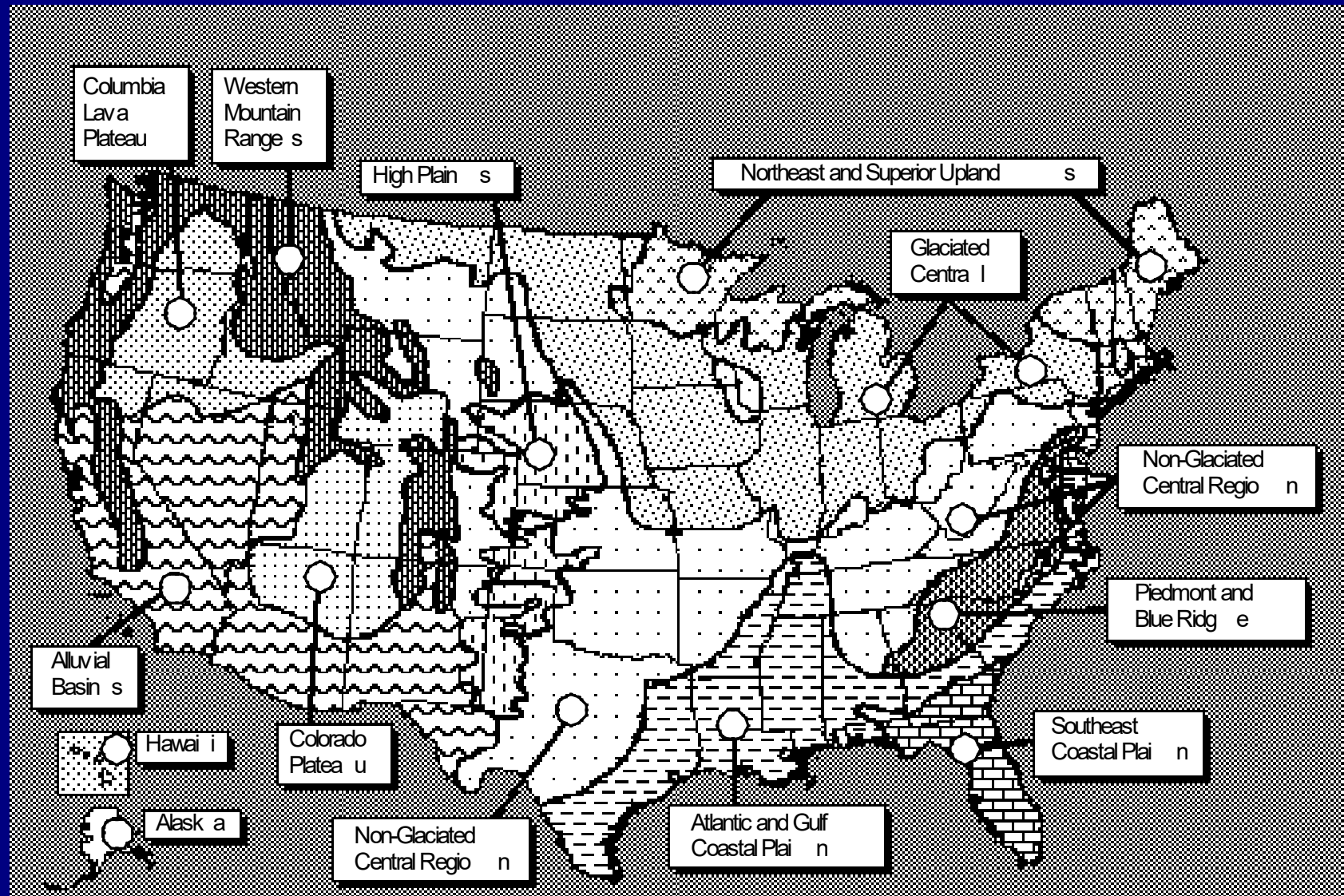
# Aquifer Systems







# U.S. Ground Water Regions

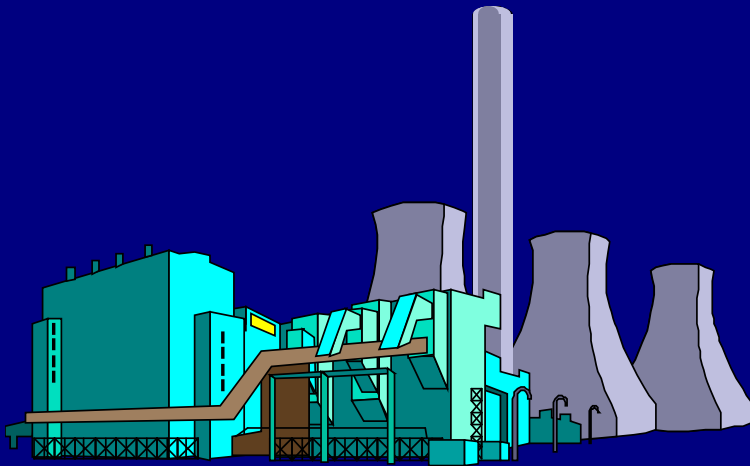


# Contamination and Remediation



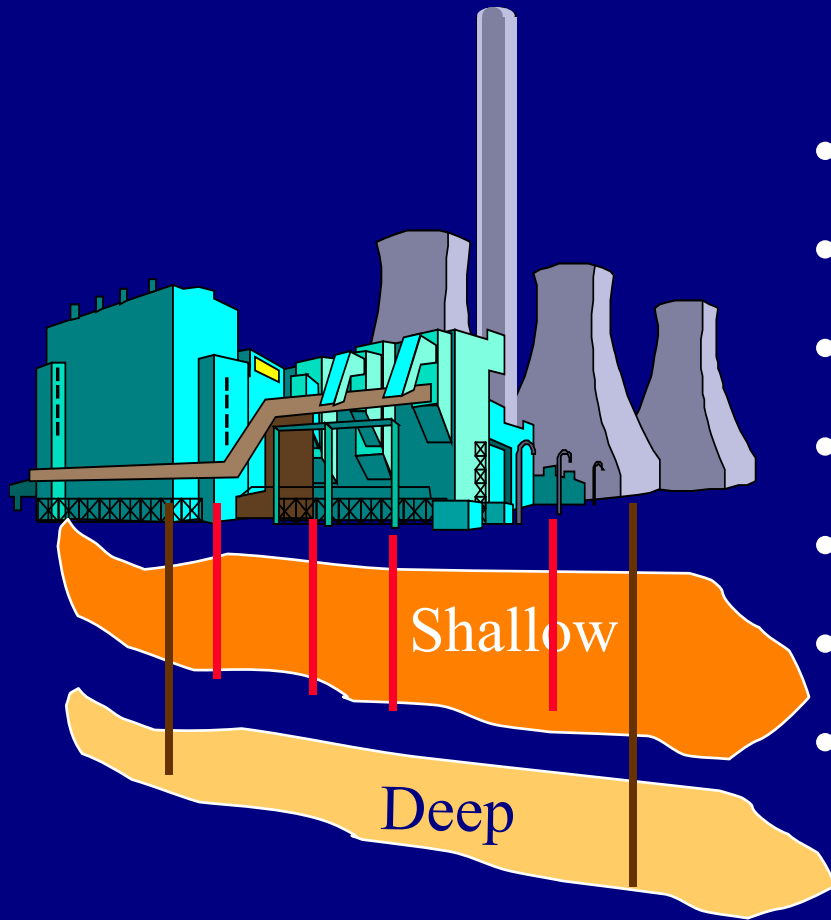
- Advective transport
- Diffusion and dispersion
- Volatilization
- Adsorption
- Biodegradation processes
- Chemical Reaction
- NAPLs
- Remediation Processes

# Sources of Contamination



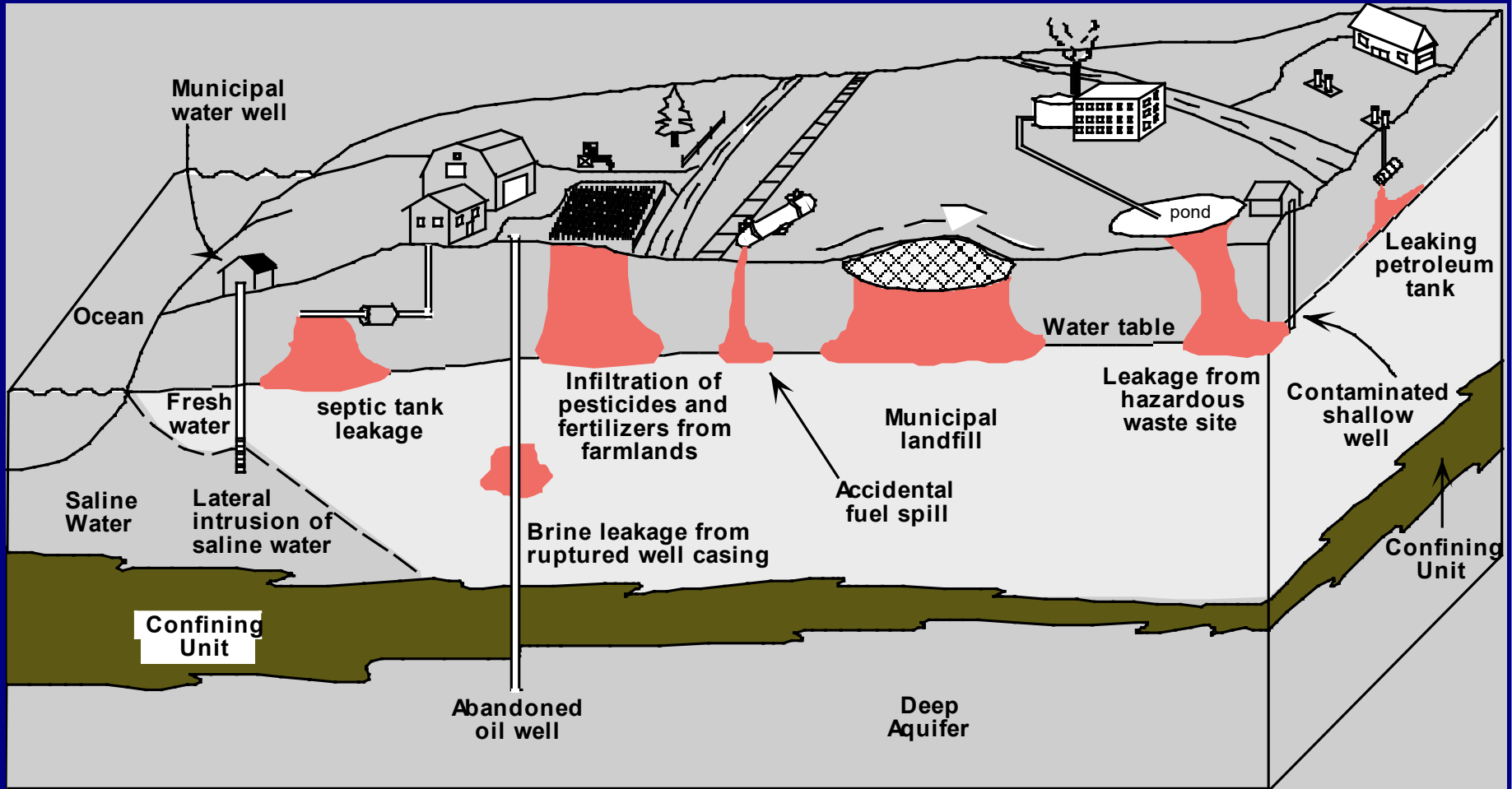
- Industrial spills and leaks
- Surface impoundments
- Storage tanks and pipes
- Landfills
- Burial areas and dumps
- Injection wells

# Areas of Industrial Contamination

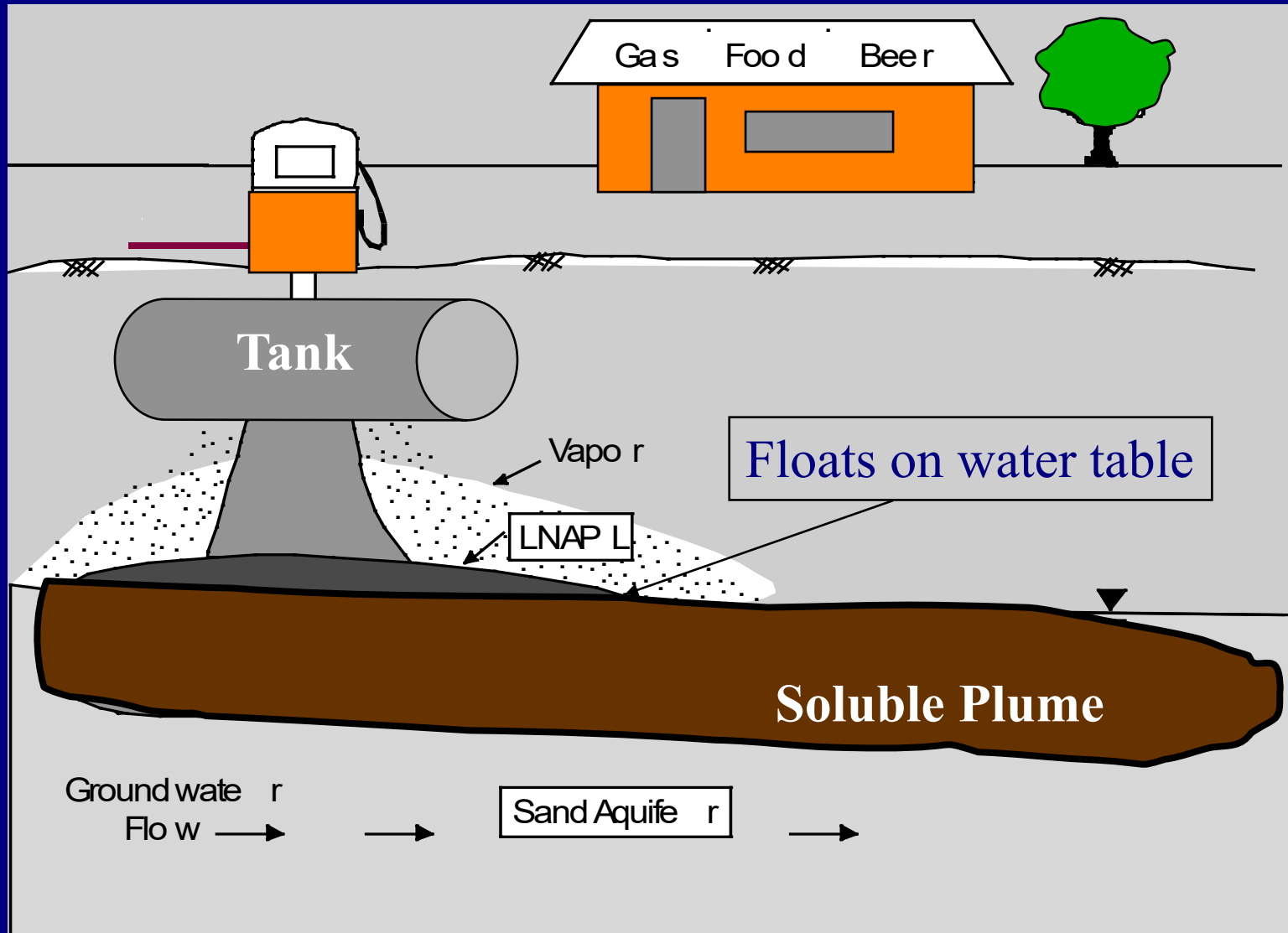


- **Surface soils**
- **Subsurface soils**
- **Shallow ground water**
- **Deep ground water**
- **Vapors above water table**
- **Drinking water wells**
- **Receiving streams/lakes**

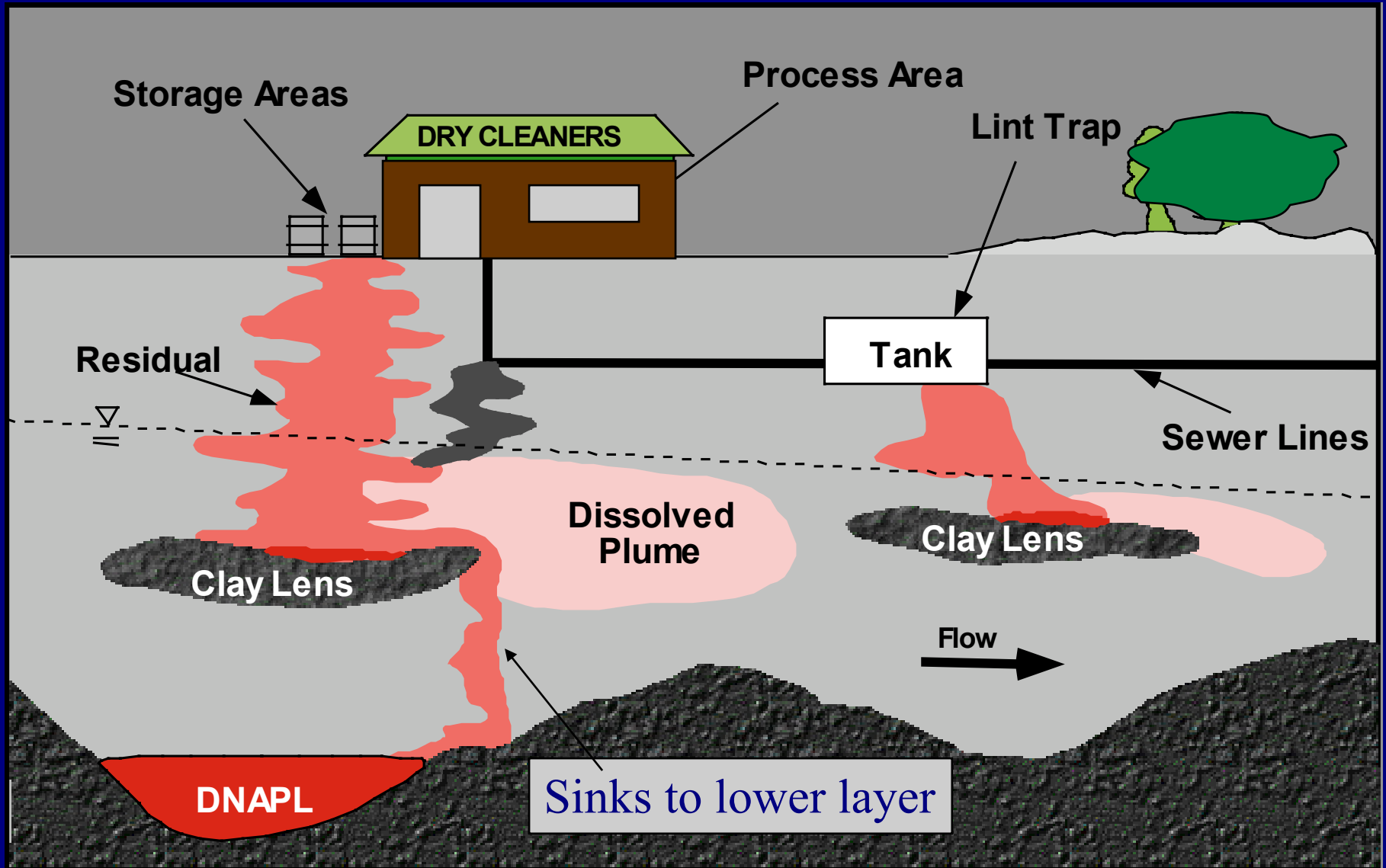
# Contamination of Ground Water



# Typical Leaking UST - BTEX

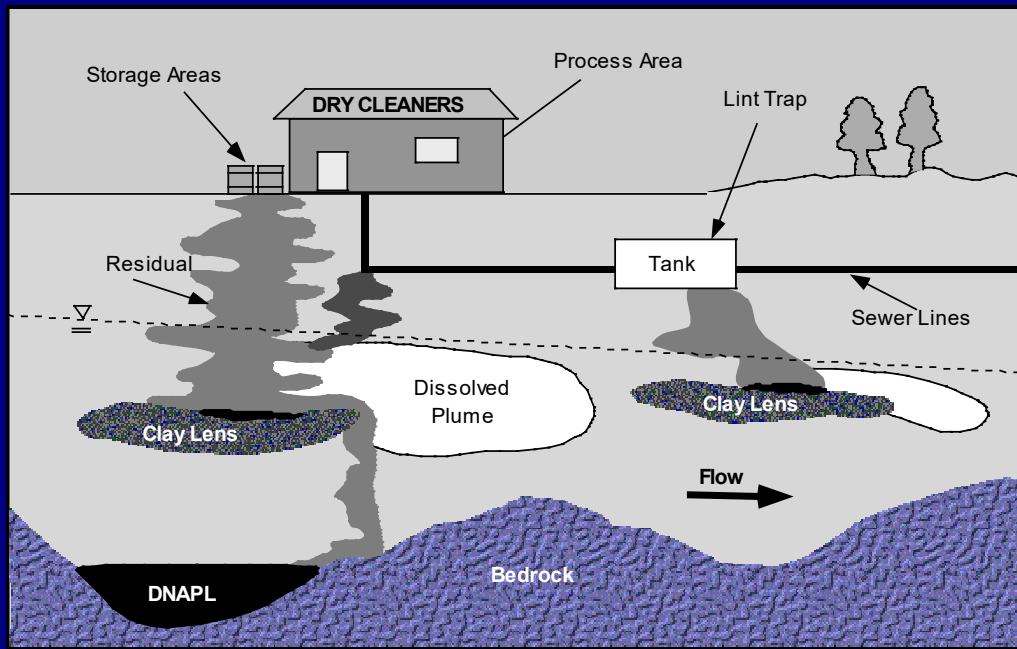


# Typical DNAPL Spill Zone



# DNAPL

## Our Most Difficult Challenge



- DNAPL source
- Residual phase
- Trapped on lenses
- Pools in low areas
- Creates soluble plumes for years
- Extremely hard to remediate

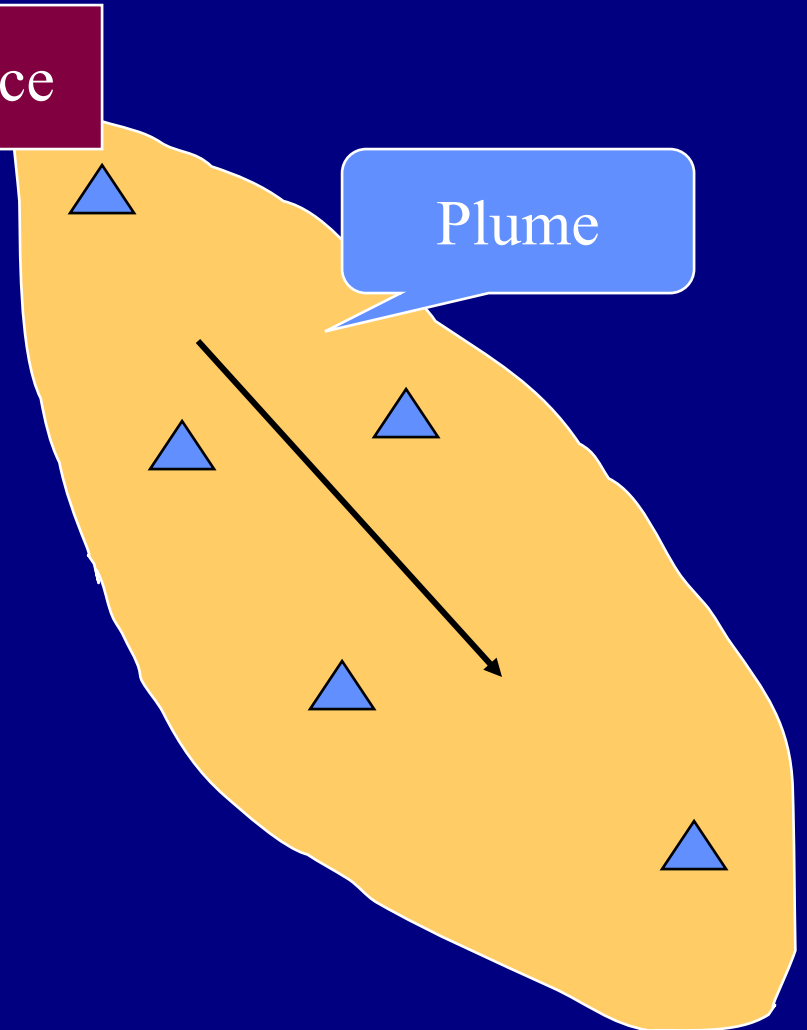


# Typical Industrial Site

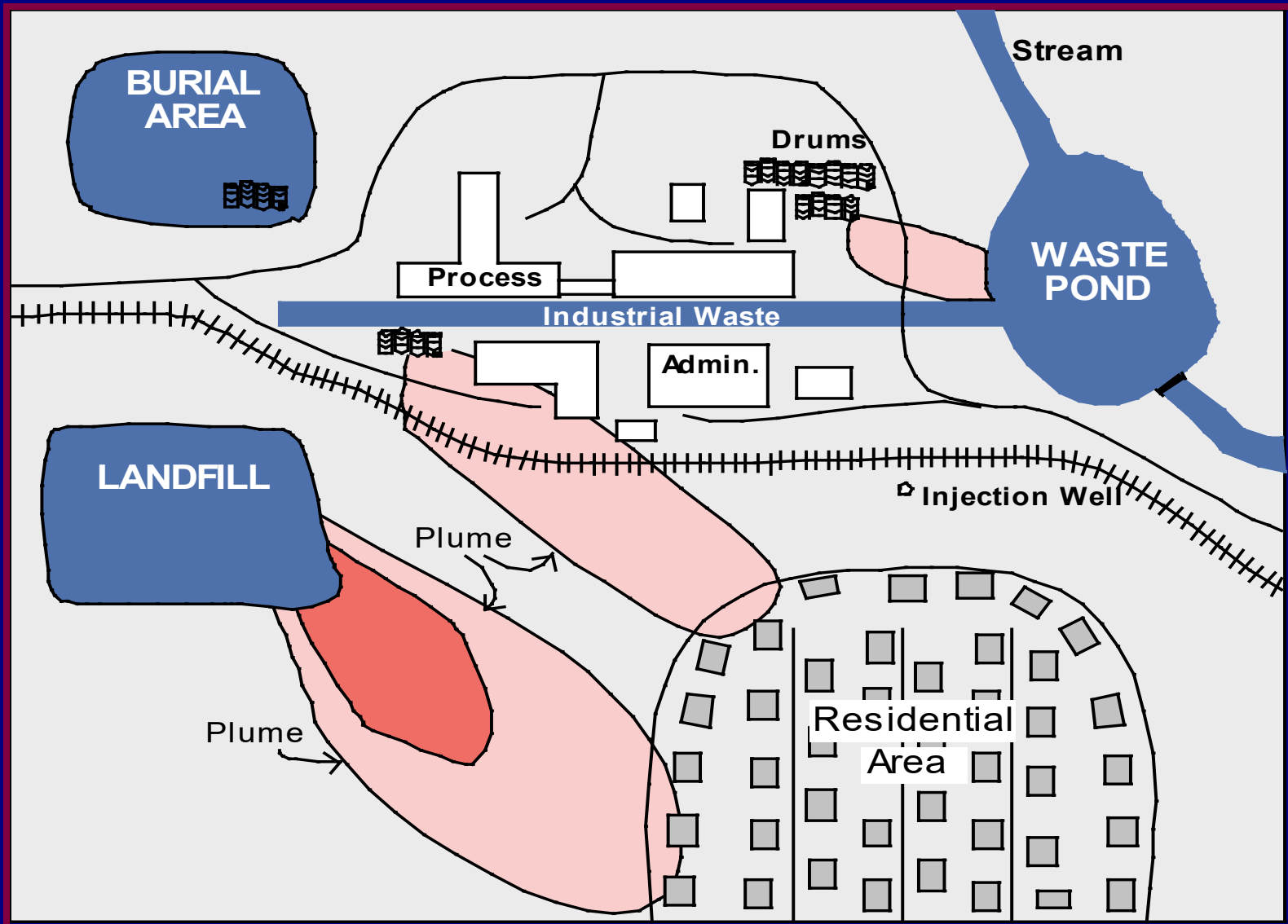
Source

Plume

- Buried fuel tanks
- Above ground chem tanks
- Ponds and Impoundments
- Buried drums (older)
- Landfill area (hidden)
- Waste process area
- Receiving streams/lakes
- Nearby residential area



# Typical Contaminated Site



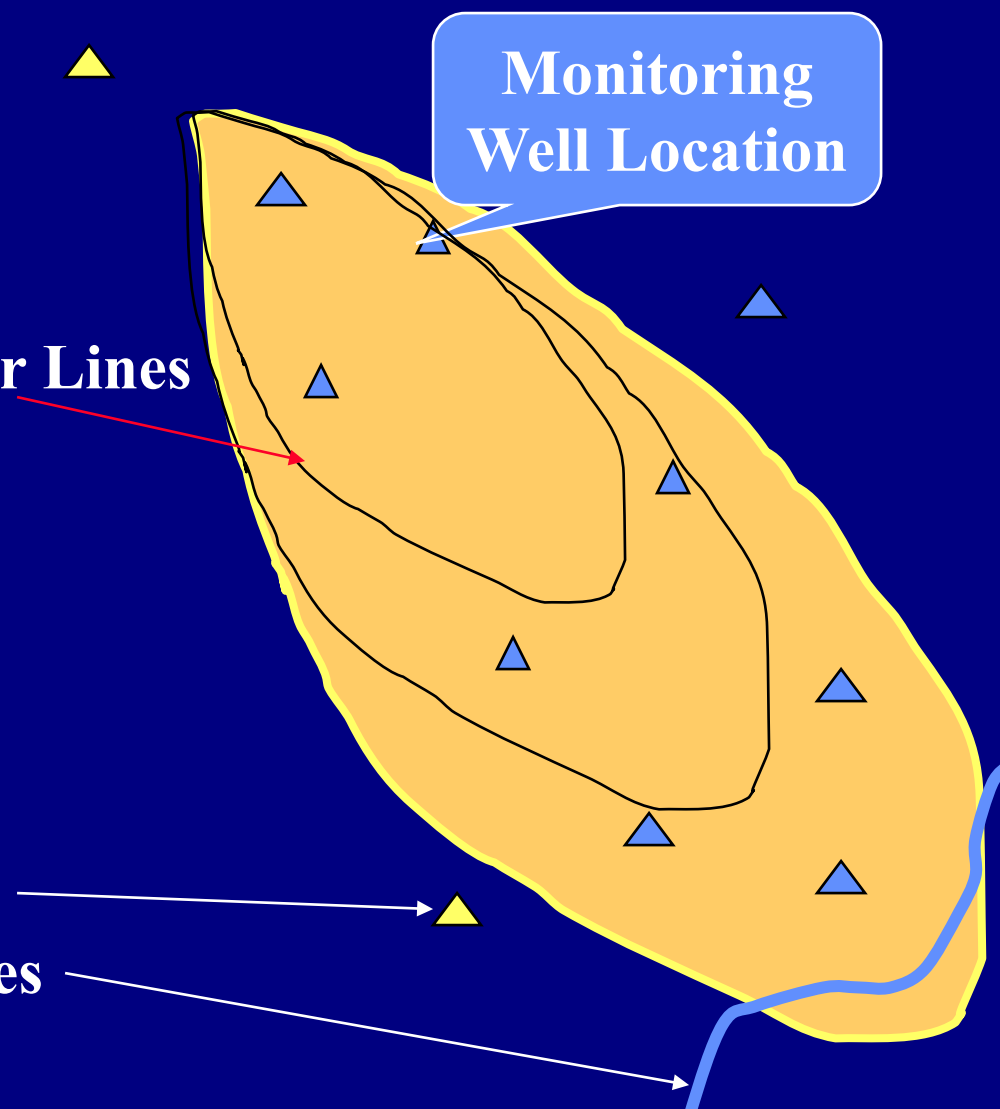
# Objectives of a Field Site Study

- **Evaluate:**

- Surface soils
- Subsurface soils
- Shallow ground water
- Deep ground water
- Vapors in subsurface
- Drinking water wells
- Receiving streams/lakes

Contour Lines

Monitoring Well Location



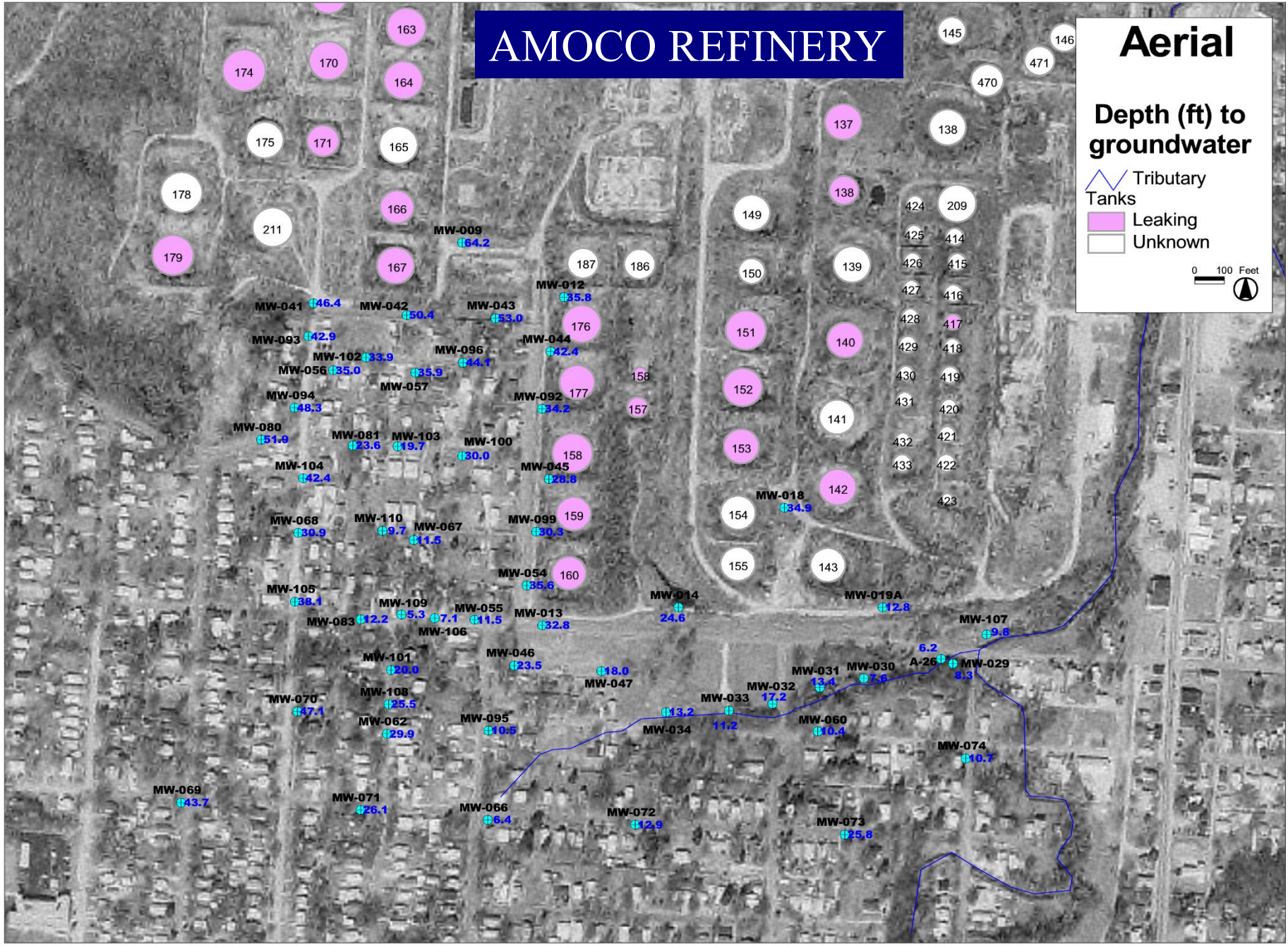
# AMOCO REFINERY

## Aerial

**Depth (ft) to groundwater**

-  Tributary
-  Tanks
-  Leaking
-  Unknown

0 100 Feet



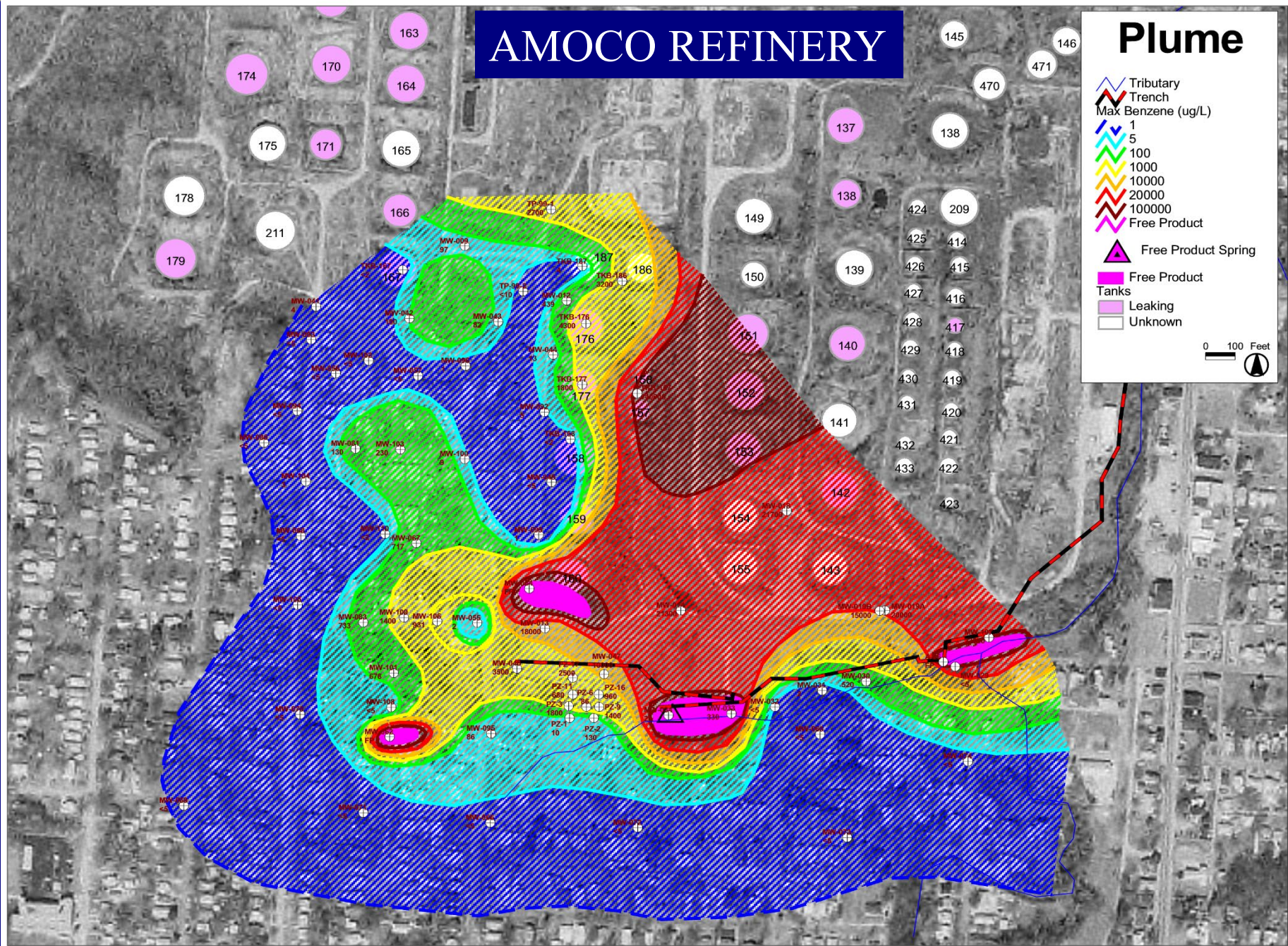


# AMOCO REFINERY

## Plume

- Tributary
- Trench
- Max Benzene (ug/L)**
- 1
- 5
- 100
- 1000
- 10000
- 20000
- 100000
- Free Product
- Free Product Spring
- Free Product Tanks
- Leaking
- Unknown

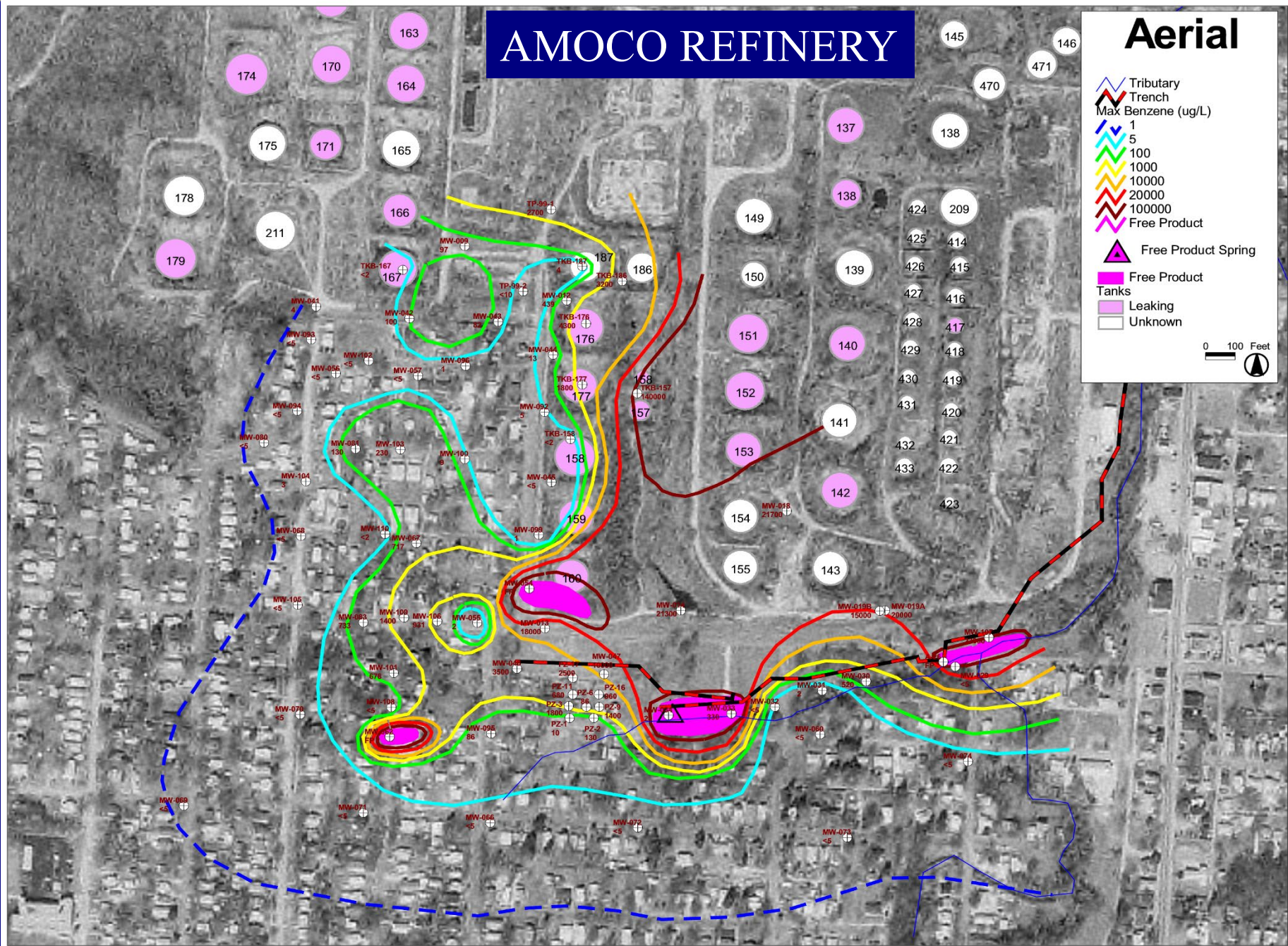
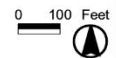
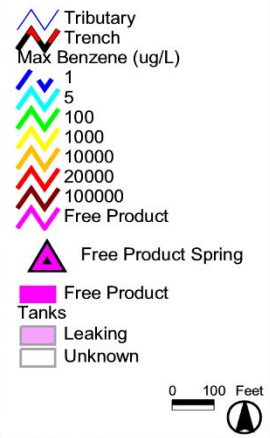
0 100 Feet





# AMOCO REFINERY

## Aerial

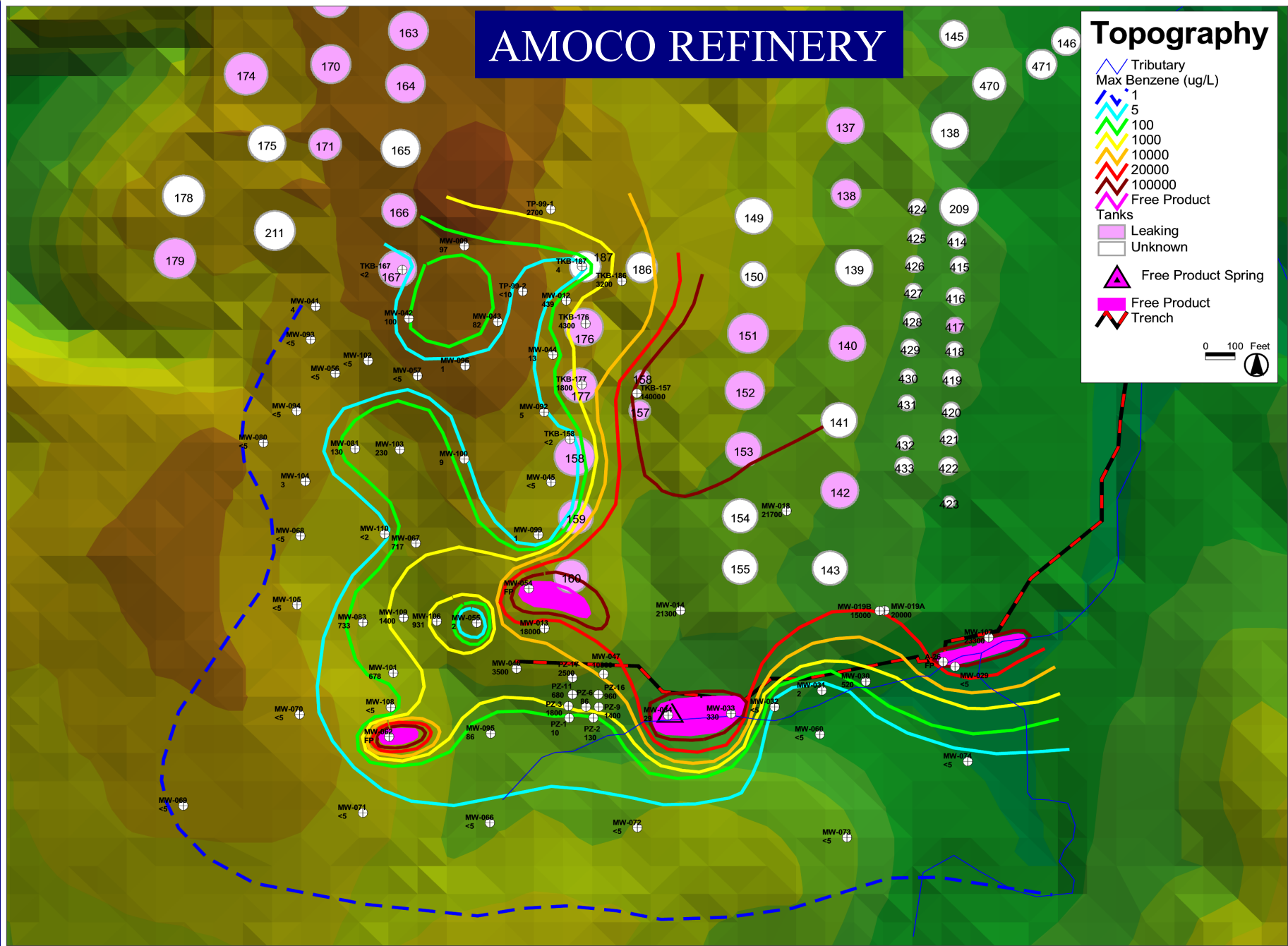


# AMOCO REFINERY

## Topography

- Tributary
- Max Benzene (ug/L)**
  - 1
  - 5
  - 100
  - 1000
  - 10000
  - 20000
  - 100000
  - Free Product
- Tanks**
  - Leaking
  - Unknown
- Free Product Spring
- Free Product
- Trench

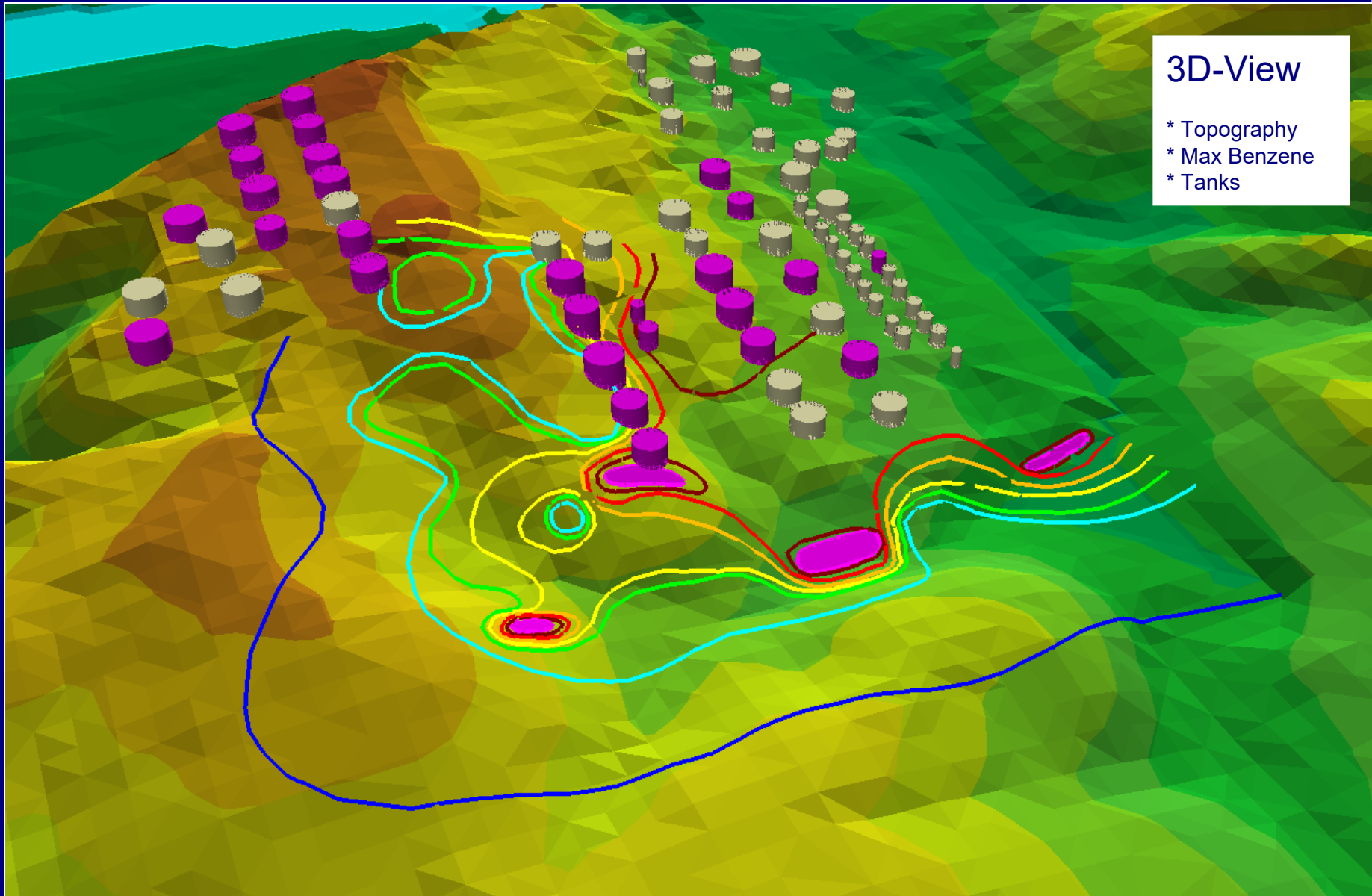
0 100 Feet





**3D-View**

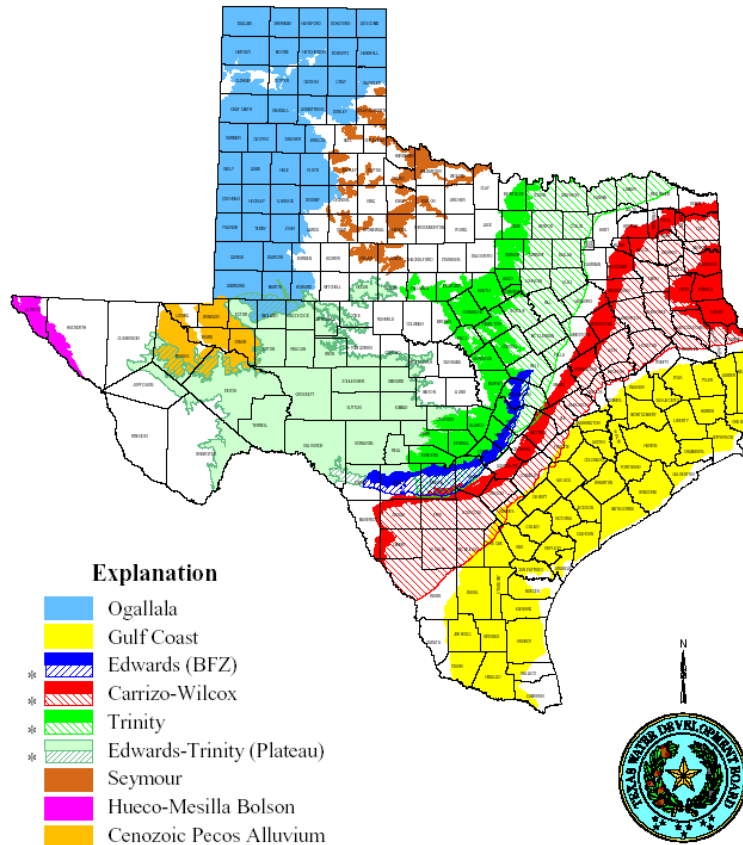
- \* Topography
- \* Max Benzene
- \* Tanks



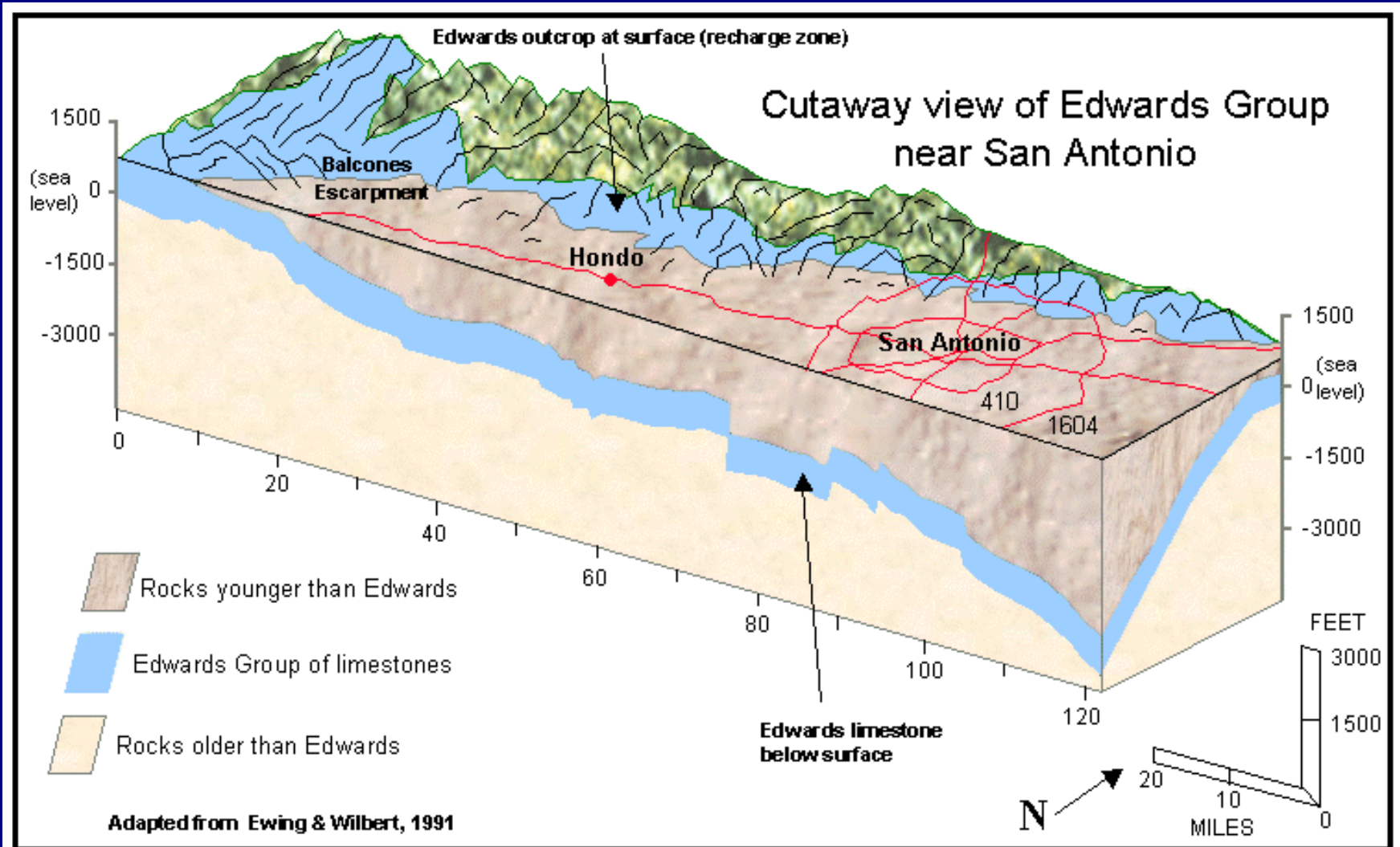


# The Major Aquifers of Texas

Major Aquifers of Texas



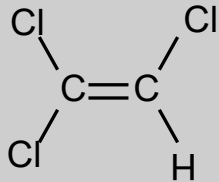
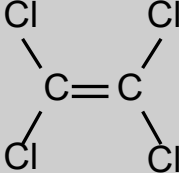
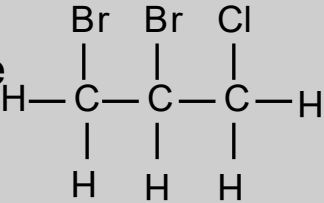
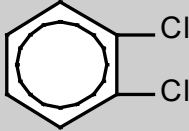
# The Edwards Group




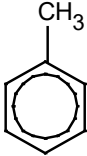
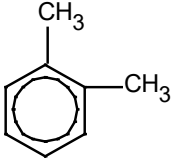
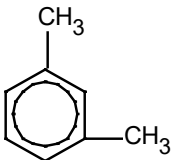
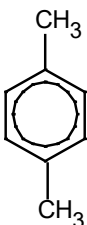
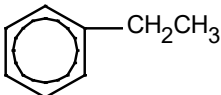
# Chlorinated Solvents

Name	Structure	Uses and Other Sources
Trichloromethane (chloroform)	$\begin{array}{c} \text{Cl} \\   \\ \text{Cl}-\text{C}-\text{Cl} \\   \\ \text{H} \end{array}$	Liquid used in manufacture of anesthetics, pharmaceuticals, fluorocarbon refrigerants and plastics. Used as solvent and insecticide. Formed from methane when chlorinating drinking water.
Vinyl chloride (chloroethene)	$\begin{array}{c} \text{H} & & \text{H} \\ & \diagdown & / \\ & \text{C}=\text{C} & \\ & / & \diagdown \\ \text{H} & & \text{Cl} \end{array}$	Gas used in the manufacture of polyvinyl chloride. End product of microbial degradation of chlorinated ethenes.
Chloroethane	$\begin{array}{c} \text{H} & \text{H} \\   &   \\ \text{H}-\text{C}-\text{C}-\text{Cl} \\   &   \\ \text{H} & \text{H} \end{array}$	Liquid used to manufacture tetraethyl lead. Degradation product of chlorinated ethanes.
1,2-Dichloroethane	$\begin{array}{c} \text{H} & \text{H} \\   &   \\ \text{Cl}-\text{C}-\text{C}-\text{Cl} \\   &   \\ \text{H} & \text{H} \end{array}$	Liquid used to manufacture vinyl chloride. Degradation product of trichloroethane.

# Chlorinated Solvents (con't)

Name	Structure	Uses and Other Sources
<u>Trichloroethene</u> (Trichloroethylene)		Solvent used in dry cleaning and metal degreasing. Organic synthesis. Degradation product of tetrachloroethene.
<u>Tetrachloroethene</u> (perchloroethene) (perchloroethylene)		Solvent used in dry cleaning and metal degreasing. Used to remove soot from industrial boilers. Used in manufacture of paint removers and printing inks.
1,2-Dibromo-3-chloropropane (DBCP)		Soil fumigant to kill nematodes. Intermediate in organic synthesis.
o-Dichlorobenzene (1,2-dichlorobenzene)		Chemical intermediate. Solvent. Fumigant and insecticide. Used for industrial odor control. Found in sewage form odor control chemicals used in toilets.

# BTEX-Related Compounds

Name	Structure	Molecular Weight	Solubility in Water	Soil-Water Partition Coefficient
Benzene		78.11	<u>1780 mg/L</u>	97
Toluene		92.1	500 mg/L	242
Xylene, ortho		106.17	170 mg/L	363
Xylene, meta		106.17	173 mg/L	182
Xylene, para		106.17	200 mg/L	331
Ethyl benzene		106.17	150 mg/L	622