

Contaminant Transport Equations

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Transport

Advection

- The process by which solutes are transported by the bulk of motion of the flowing ground water.
- Nonreactive solutes are carried at an average rate equal to the average linear velocity of the water.

Hydrodynamic Dispersion

- Tendency of the solute to spread out from the advective pathway
- Two processes
 - Diffusion (molecular and turbulent)
 - Dispersion (velocity differences in space)

Diffusion

- Ions (molecular constituents) in solution move under the influence of kinetic activity in direction of their concentration gradients.
- Occurs in the absence of any bulk hydraulic movement
- Diffusive flux is proportional to concentration gradient, per *Fick's First Law* ($M L^{-2} T^{-1}$).

$$F = -D_m \left(\frac{dc}{dx} \right)$$

- Where D_m = diffusion coefficient (typically 1×10^{-5} to 2×10^{-5} cm^2/s for major ions in ground water)

Diffusion (continued)

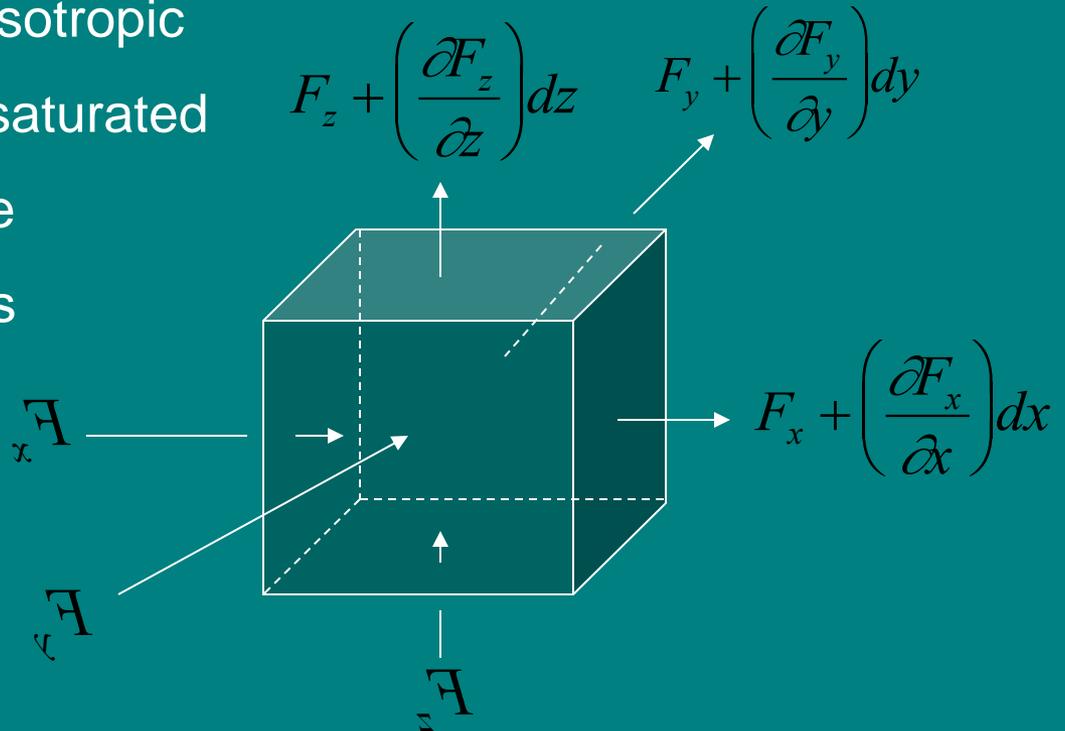
- ***Fick's Second Law*** - derived from Fick's First Law and the Continuity Equation - called "Diffusion Equation"

$$\frac{\partial C}{\partial t} = D_m \left(\frac{\partial^2 C}{\partial x^2} \right)$$

The Advection-Dispersion Equation Derivation (F is transport flux)

Assumptions:

- 1) Porous medium is homogenous
- 2) Porous medium is isotropic
- 3) Porous medium is saturated
- 4) Flow is steady-state
- 5) Darcy's Law applies



Advection Dispersion Equation

In the x-direction:

$$\text{Transport by advection} = \bar{v}_x n C dA \quad \text{units} \frac{M}{T}$$

$$\text{Transport by dispersion} = n D_x \left(\frac{\partial C}{\partial x} \right) dA \quad \text{units} \frac{M}{T}$$

Where:

\bar{v} = average linear velocity

n = porosity (constant for unit of volume)

C = concentration of solute

dA = elemental cross-sectional area of cubic element

$$D_x = \alpha_x \bar{v}_x + D_m$$

Hydrodynamic Dispersion D_x caused by molecular diffusion and variations in the velocity field and heterogeneities

where:

α_x = dispersivity [L]

D_m = Molecular diffusion

$$F_x = v_x nC - nD_x \left(\frac{\partial C}{\partial x} \right)$$

- Flux = (mass/area/time)

(-) sign before dispersion term indicates that the contaminant moves toward lower concentrations

$$F_x = v_x nC - nD_x \left(\frac{\partial C}{\partial x} \right)$$

- Total amount of solute entering the cubic element

$$= F_x dydz + F_y dxdz + F_z dxdy$$

- Difference in amount entering and leaving element =

$$\left(\frac{\partial F_x}{\partial x} + \frac{\partial F_y}{\partial y} + \frac{\partial F_z}{\partial z} \right) dx dy dz$$

- For nonreactive solute, difference between flux in and out = amount accumulated within element
- Rate of mass change in element =

$$n \left(\frac{\partial C}{\partial t} \right) dx dy dz$$

- Equate two equations and divide by $dV = dx dy dz$:

$$-\left(\frac{\partial F_x}{\partial x}\right) + \left(\frac{\partial F_y}{\partial y}\right) + \left(\frac{\partial F_z}{\partial z}\right) = n \left(\frac{\partial \mathcal{C}}{\partial t}\right)$$

- Substitute for fluxes and cancel n:

$$-\left[\frac{\partial}{\partial x}(\bar{v}_x C) + \frac{\partial}{\partial y}(\bar{v}_y C) + \frac{\partial}{\partial z}(\bar{v}_z C)\right] +$$

$$\left\{\frac{\partial}{\partial x}\left[D_x\left(\frac{\partial \mathcal{C}}{\partial x}\right)\right] + \frac{\partial}{\partial y}\left[D_y\left(\frac{\partial \mathcal{C}}{\partial y}\right)\right] + \frac{\partial}{\partial z}\left[D_z\left(\frac{\partial \mathcal{C}}{\partial z}\right)\right]\right\} = \frac{\partial \mathcal{C}}{\partial t}$$

- For a homogenous and isotropic medium, \bar{v} is steady and uniform.

- For a homogenous and isotropic medium, D is steady and uniform.
- Therefore, D_x , D_y , and D_z do not vary through space.

$$\left[D_x \left(\frac{\partial^2 C}{\partial x^2} \right) + D_y \left(\frac{\partial^2 C}{\partial y^2} \right) + D_z \left(\frac{\partial^2 C}{\partial z^2} \right) \right]$$

- **Advection-Dispersion Equation 3-D:**

$$\left[\bar{v}_x \left(\frac{\partial C}{\partial x} \right) + \bar{v}_y \left(\frac{\partial C}{\partial y} \right) + \bar{v}_z \left(\frac{\partial C}{\partial z} \right) \right] = \frac{\partial C}{\partial t}$$

In 1-D, the AD equation thus becomes:

$$\frac{\partial C}{\partial t} + \bar{v}_x \left(\frac{\partial C}{\partial x} \right) = D_x \left(\frac{\partial^2 C}{\partial x^2} \right)$$

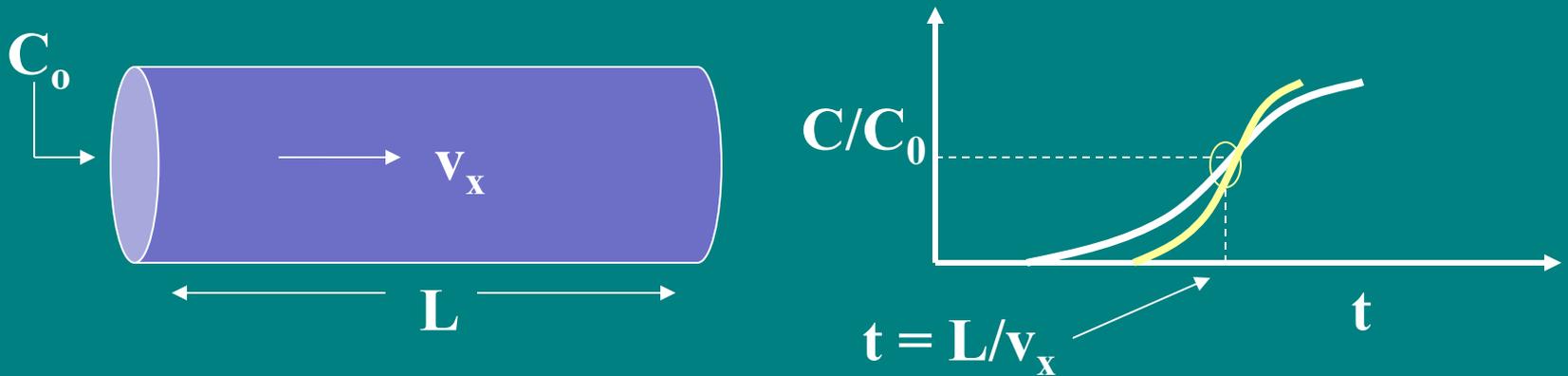
↑
Accumulation

↑
Advection

↑
Dispersion

CONTINUOUS SOURCE

- Solution for 1-D Equation for can be found using Laplace Transform



- 1-D soil column breakthrough curves

Solution can be written (Ogato & Banks, 1961)

$$\frac{C}{C_0} = \frac{1}{2} \operatorname{Erfc} \left(\frac{L - v_x t}{\sqrt{4D_x t}} \right) + \left(\frac{1}{2} \right) \operatorname{Exp} \left(\frac{v_x L}{D_x} \right) \operatorname{Erfc} \left(\frac{L - v_x t}{\sqrt{4D_x t}} \right)$$

or, in most cases

$$\frac{C}{C_0} = \frac{1}{2} \operatorname{Erfc} \left(\frac{L - v_x t}{\sqrt{4D_x t}} \right)$$

where

$$\operatorname{Erfc}(B) = 1 - \operatorname{Erf}(B)$$

$$\operatorname{Erf}(B) = \frac{2}{\sqrt{\pi}} \int_0^B e^{-z^2} dz$$

Tabulated error
function

Instantaneous Sources

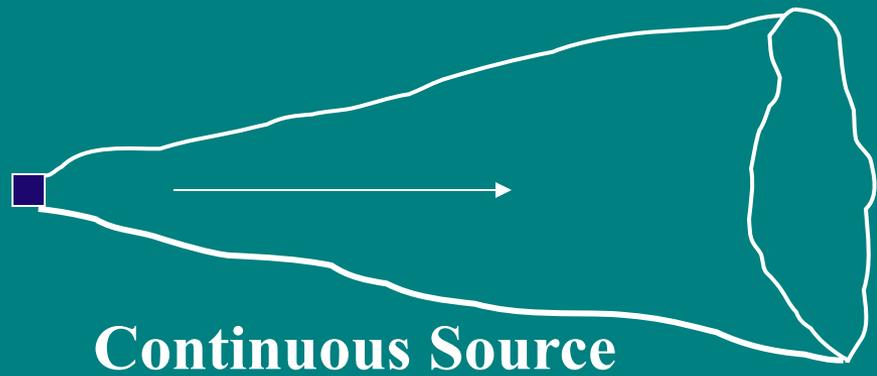
Advection-Dispersion Only

Instantaneous POINT Source 3-D:

$$C(x, y, z, t) = \frac{M}{(4\pi Dt)^{3/2}} \exp\left(-\frac{(x-vt)^2}{4D_x t} - \frac{y^2}{4D_y t} - \frac{z^2}{4D_z t}\right)$$

$$M = C_0 V$$

$$D = (D_x D_y D_z)^{1/3}$$



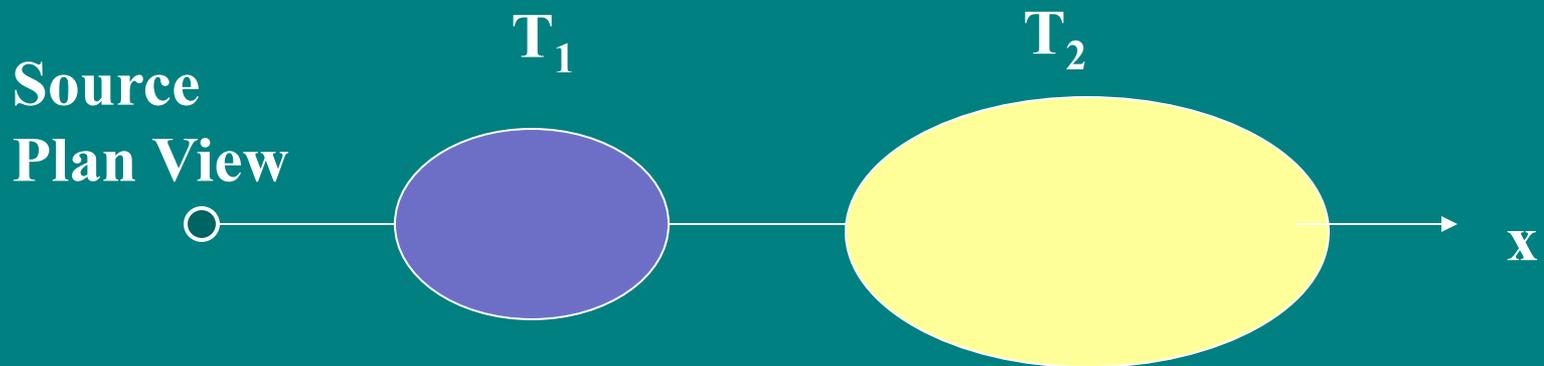
Instantaneous LINE Source 2-D Well: (with First Order Decay)

$$C(x, y, t) = \frac{m}{4\pi Dt} \exp\left(-\frac{(x - vt)^2}{4D_x t} - \frac{y^2}{4D_y t} - kt\right)$$

$k = \text{first order}$
 $\text{decay } (t^{-1})$

$$m = C_o A$$

$$D = \sqrt{D_x D_y}$$

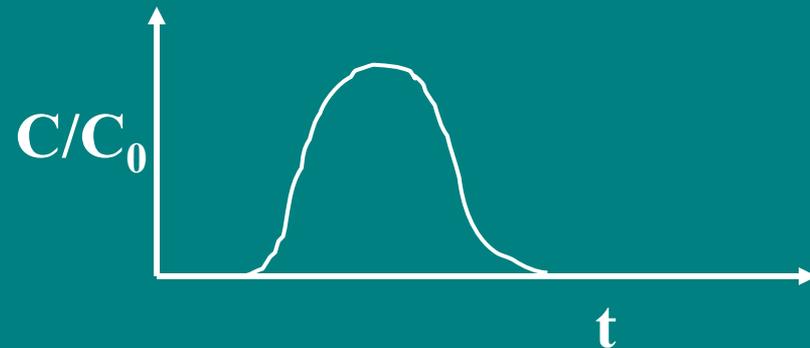
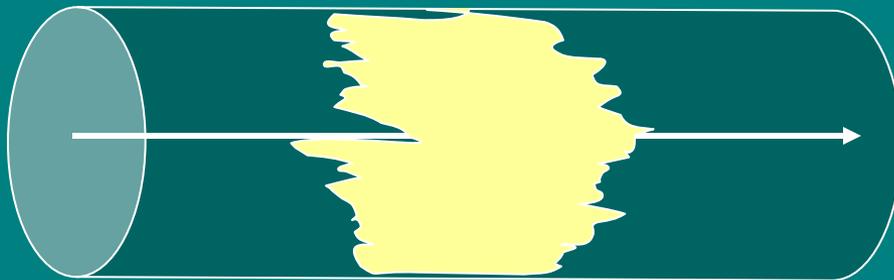


Instantaneous PLANE Source - 1 D

$$C(x, t) = \frac{M}{\sqrt{4\pi D_x t}} \text{Exp}\left(-\frac{(x - vt)^2}{4D_x t}\right) \quad M = \frac{\text{mass}}{\text{area}}$$

AD Equation

$$\frac{\partial C}{\partial t} + \bar{v}_x \left(\frac{\partial C}{\partial x} \right) = D_x \left(\frac{\partial^2 C}{\partial x^2} \right)$$



$$T = L/v_x$$